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Solution The general solution method is best explained in terms of the complex reaction network notation where the rate for the jth species is given : $N_k j_1 y_1 k j_2 y_2 \dots k l j y_j \dots dt l dy_j (1)$ A solution is assumed of the form: $y_j Y_j^0 e^{(2)}$ where Y_j and a are constants to be determined.

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P5-24 (d) A lower conversion is reached due to equilibrium. Also, the reverse reaction begins to overtake the forward reaction near the exit of the reactor. P5-24 (e) Individualized solution 5-50 P5-24 (f) Individualized solution P5-24 (g) Individualized solution P5-25 No solution necessary P5-26 A B C D P0 P 1. Mole Balance – Use Differential Form dX/dW

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Chapter 2: Conversion and Reactor Sizing Additional Homework Problems CDP2-A B Solution. Over what range of conversions are the plug-flow reactor and CSTR volumes identical? So that it is easier to visualize the solution, we first plot the inverse of the reaction rate versus conversion.

Elements of Chemical Reaction Engineering

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Appendix 4 Solution to Example 4-10: Three Equal-Volume CSTRs in Series 122. 5. Reaction Rate Fundamentals (Chemical Kinetics) 123. 5.1 Elementary Reactions 123. 5.1.1 Significance 123. 5.1.2 Definition 125. 5.1.3 Screening Criteria 126. 5.2 Sequences of Elementary Reactions 129. 5.2.1 Open Sequences 130. 5.2.2 Closed Sequences 130