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Introduction NMR

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Introduction to Nuclear
Magnetic Resonance

*How NMR spectrometer
works How to use NMR
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NMR
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Analysis Atomic
emission spectroscopy
(AES) is a method of
chemical analysis that
uses the intensity of
light emitted from a

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flame, plasma, arc, or
spark at a particular
wavelength

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2. Presentation outline

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Coupling •Fundamental
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- NOE •Chemical shift

- NOESY •Factors

affecting chemical shift

- COSY •Interpretation

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NMR spectra ...

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INTRODUCTION OF

NMR Nuclear magnetic

resonance spectroscopy

or NMR is a powerful

analytical technique

used to characterize

organic molecules by

identifying carbon-

hydrogen frameworks

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within molecules. Two common types of NMR spectroscopy are used.....

: ^1H NMR is used to determine the type and number of H atom in a molecule. : ^{13}C NMR is used to determine the type and number of C atom in a molecule. The source of energy in NMR is radio waves, which have long wavelength. 3

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NMR Spectroscopy

Nuclear magnetic resonance spectroscopy, most commonly known as NMR spectroscopy or magnetic resonance spectroscopy, is a spectroscopic technique to observe local magnetic fields around atomic nuclei. The sample is placed in a magnetic field and the

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NMR signal is produced by excitation of the nuclei sample with radio waves into nuclear magnetic resonance, which is detected with sensitive radio receivers. The intramolecular magnetic field around an atom in a molecule changes the resonance frequen

Nuclear magnetic

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NMR spectroscopy.

Nuclear magnetic

resonance (NMR) is a

spec-troscopic technique

that detects the energy

ab-sorbed by changes in

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the nuclear spin state.

The application of NMR spectroscopy to the study of proteins and nucleic acids has provided unique information on the dynamics and chemical kinetics of these systems. One important feature of NMR

Chapter 1

INTRODUCTION TO

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NMR SPECTROSCOPY

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spectroscopy,

wavelength usually is

expressed in nanometers

(1 nm = 10^{-9} m). It

follows from the above

equations that radiation

with shorter wavelength

has higher energy. In

UV-visible

spectroscopy, the low-

wavelength UV light

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has the highest energy.

In some cases, this energy is

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NMR is like a camera with a slow shutter speed and a blurred image of fast processes is observed. At room temperature the cyclohexane protons are observed as a time-

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average and appear as a single resonance. At -90°C the chair-chair interconversion is sufficiently slow that axial and equatorial are observed as two separate resonances. 8

*Chapter 13: Nuclear
Magnetic Resonance
(NMR) Spectroscopy*

NMR spectroscopy:
Besides ^1H and ^{13}C

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many other "good" NMR nuclei (e.g., ^{11}B , ^{19}F , ^{31}P , and ^{195}Pt) give important information on compound properties and structure. Also the NMR of paramagnetic species can result in important structural information. Proton NMR is also important because the light hydrogen nucleus is not

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easily detected by X...

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