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 $1/8 = 1$ atom 6
face atoms $\times 1/2 =$
3 atoms Total of
4 atoms per unit
cell (b) bcc: 8
corner atoms \times
 $1/8 = 1$ atom 1
enclosed atom =
1 atom Total of
2 atoms per unit
cell (c)

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Diamond: 8

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 $1/8 = 1$ atom 6

face atoms $\times 1/2 =$

3 atoms 4

enclosed atoms =

4 atoms Total of

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Chapter 3 3.1 If
a ϕ were to
increase, the
bandgap energy
would decrease
and the material
would begin to

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less like a semiconductor and more like a metal. If α_0 were to decrease, the bandgap energy would increase and the material would begin to behave more like an insulator.

3.2 wave

equation is: $\frac{1}{2} \rho v^2 A^2$

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