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Types of Chemical Reactions Introduction to Balancing Chemical Equations Worksheet 6 2 Chemical Reaction

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oxide powder to produce aluminum oxide solid and iron metal.



## Worksheet 6.2 Word Equations

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Writing and Balancing Equations 1. Write the chemical equations and balance each of the following word equations. a) Aluminum metal reacts with iron (II) oxide powder to produce aluminum oxide solid and

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Worksheet 6-2 Chemical Reaction Types Name Period Identify each chemical reaction as a synthesis (combination), decomposition,

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single- replacement, double-replacement, or combustion reaction. 1, 2, 3. 4. 5. 6, 8. 9. 10.  $2\text{S}\text{O}_2 + \text{O}_2 \rightarrow 2\text{S}\text{O}_3$   $3\text{CaS}\text{O}_4 + 5\text{O}_2 \rightarrow 2\text{H}_2\text{O}$   $\text{Mg} + 2\text{AgN}\text{O}_3 \rightarrow 2\text{Ag} + 2\text{H}_3\text{P}\text{O}_4 + 6\text{HN}\text{O}_3$   $\text{MgCl}_2 + 3\text{O}_2 \rightarrow 2\text{Be} + \text{O}_2$   $2\text{BeO}$   $2\text{Al} + 3\text{Cu} \rightarrow 2\text{PbO}_2$   $2\text{PbO} + \text{O}_2$

Ms. Chab's Website - Home

Section Quick Check Date CHAPTER 6 Section 2: Chemical Reactions Class After reading the section in your textbook, respond to each statement. 1. State the term for the amount of energy that is needed for a chemical reaction to occur. 2. Summarize the relationship between an enzyme and a substrate. enc 3. <https://www.svcsd.org/cms/lib07/NY01913388/Centricity/Domain/245/biochem%20review.pdf>.

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2 P<sub>2</sub> O<sub>3</sub> synthesis 13. 2 RbNO<sub>3</sub> + BeF<sub>2</sub> ... Types of Chemical  
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hbr type of reaction double displacement 2 3 caoh

Six Types Of Chemical Reaction Worksheet Answers [PDF]

This is the chemical reaction where an oxygen compound and carbon compound combine together to become  $H_2O$  and  $CO_2$ . It is the reaction where mostly an organic compound like oxygen burns yielding to water, carbon dioxide, or some other product. The combination of any substance with oxygen results in combustion.

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Chemical Creations Worksheet / Worksheet (teacher made)

Atoms are rearranged during chemical reactions, and are not lost or gained. Chemical reactions can be represented using equations. Catalysts speed up reactions without being used up.

Chemical reactions - Types of reaction - KS3 Chemistry ...  
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available in our digital library an ... Numbers Chapter 6 2. Chemical  
reactions A chemical change always involves a rearrangement of  
atoms and the expression of the atomic rearrangement like this is  
called a chemical reaction.

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